- Boiling Point
  - vapor pressure of liquid = atmospheric pressure
  - normal boiling point: temp. at 760 mm Hg
  - ↓ Pressure, ↓ temp. of boiling point

- Critical Temp. & Critical Pressure
  - critical point: where the liquid reached critical temperature (high enough temp.) & critical pressure (high enough pressure)
    - supercritical fluid

- Surface Tension: energy required to break through the surface
  - water drops to be spread

- Capillary action: rising of water due to attraction of two substances
  - meniscus
    - adhesive forces: attraction between liquid & non-liquid (solid)
    - cohesive forces: water molecules themselves

- Viscosity: resistance of liquids to flow
  - highly viscous: long chains of molecules in greater intermolecular forces