

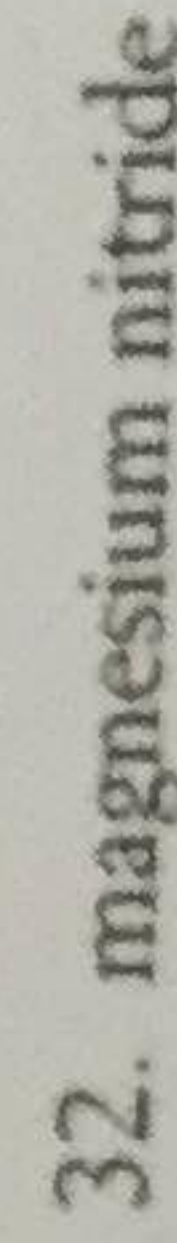
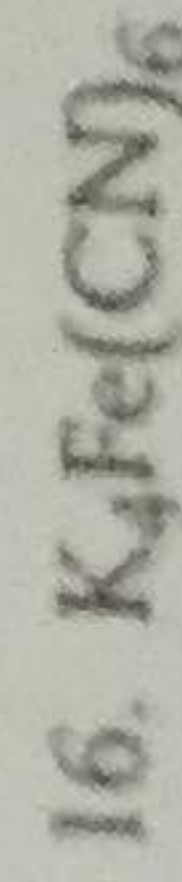
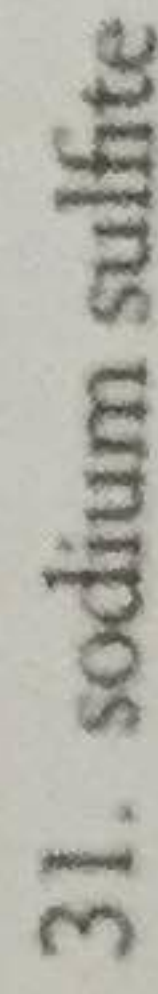
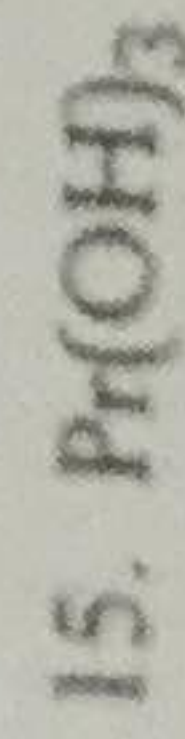
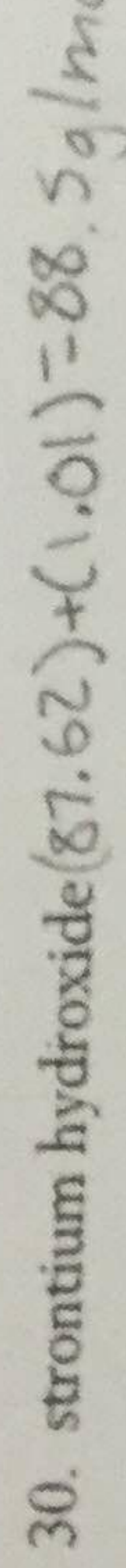
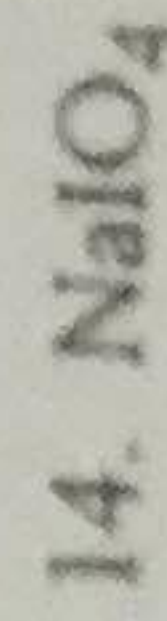
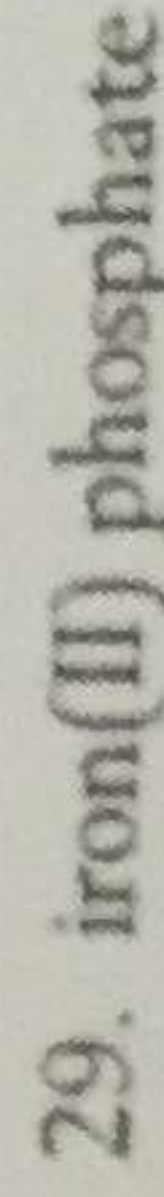
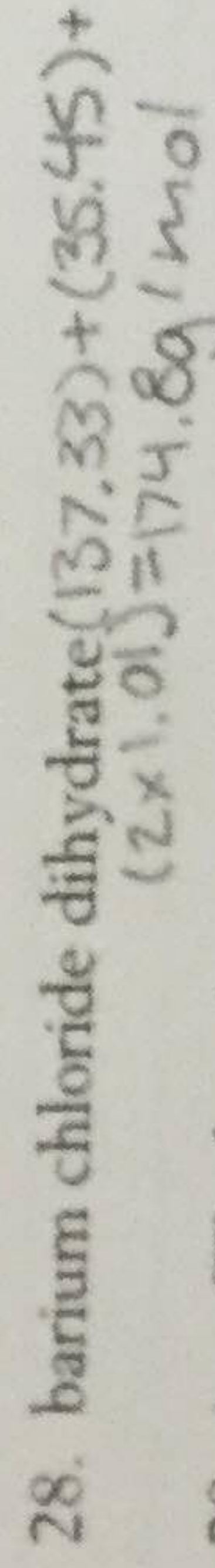
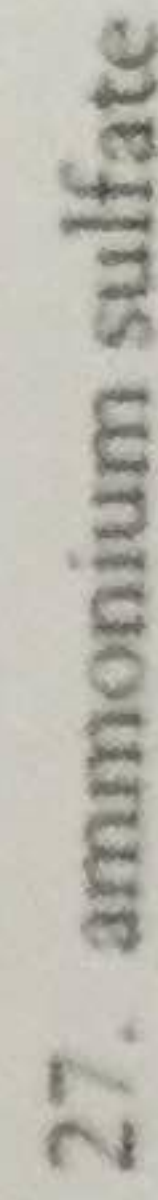
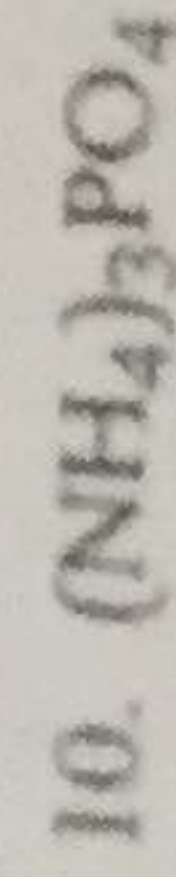
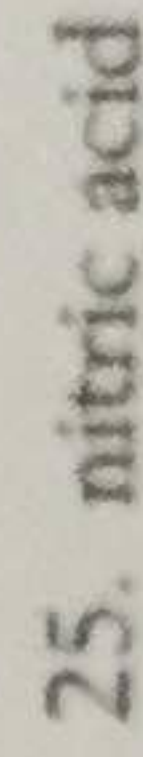
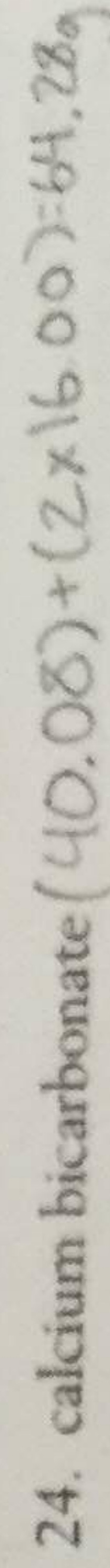
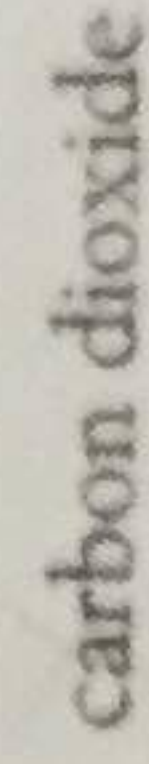
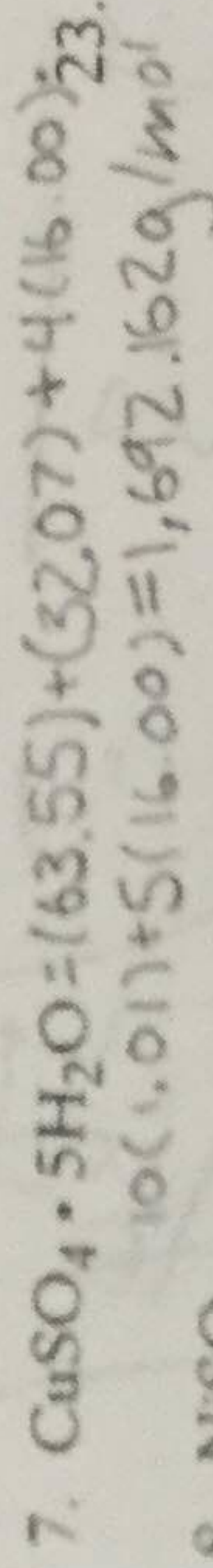
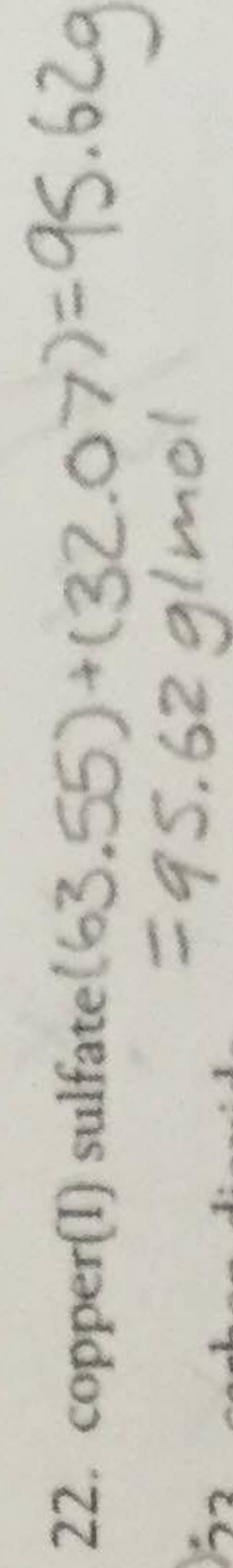
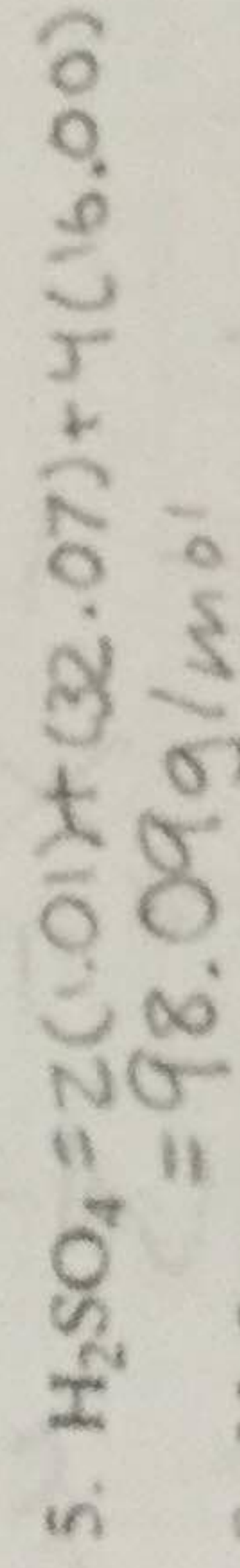
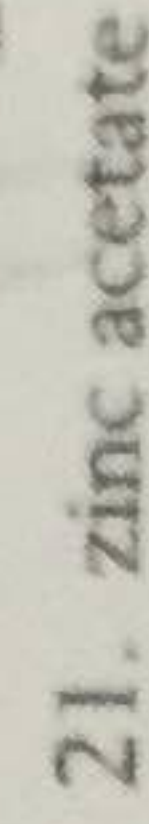
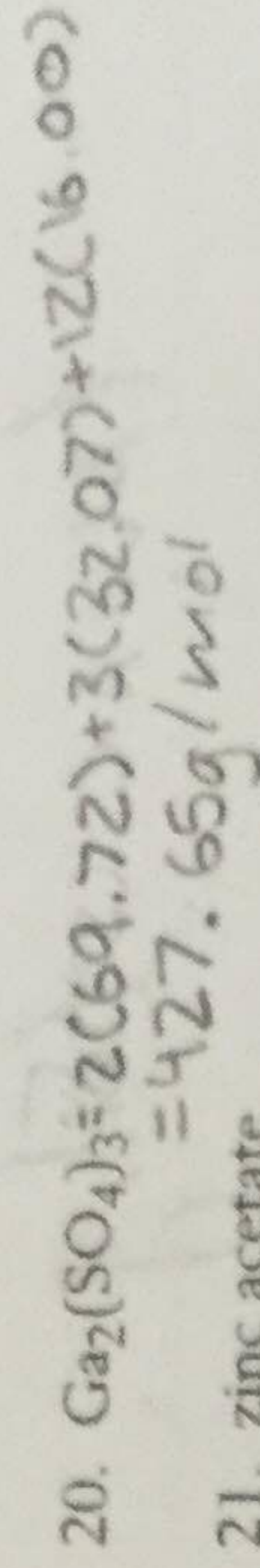
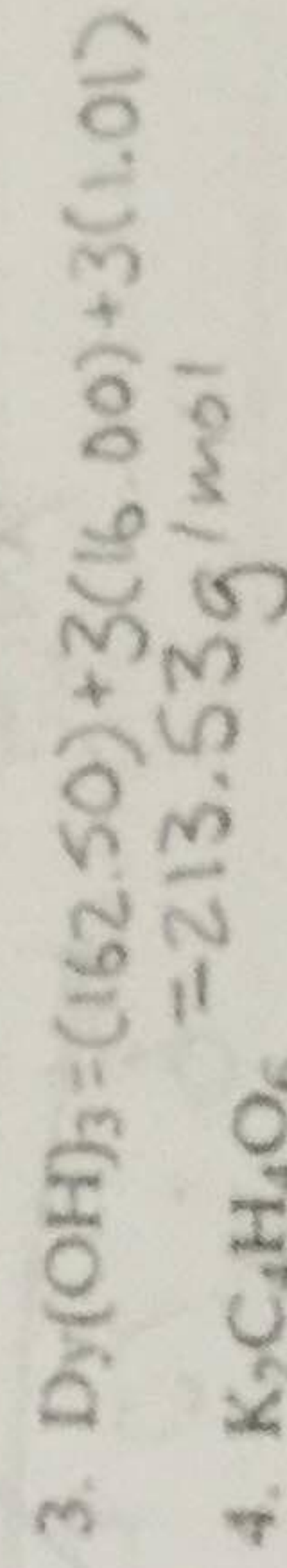
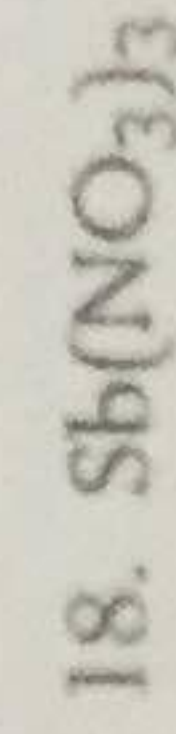
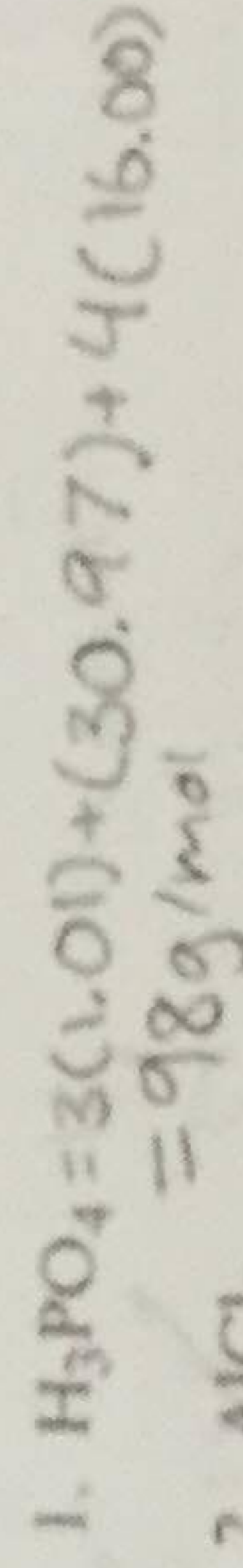


Molecular Mass and Mole Calculations

Name Esbeyda Del Rio

Date 2/3/16

Group 1. Find the molecular mass or formula mass for each of the compounds shown below.



(continued)



Mixtures

Mixtures

Any two or more things combined: ex - Air, brass, salt water, lemonade, salads, gatorade, kool-aid, etc. Today we learned

we took a short set notes on Mixtures. We learned

Homogeneous

Heterogeneous

what's the difference between some homogeneous and heterogeneous mixtures? We also learned what homogeneous and heterogeneous mixtures are.

Homogeneous - uniformly distributed
Heterogeneous - not uniformly distributed

Heterogeneous-mix (micible)

homogeneous mix



micible - mixes all around / immicible - doesn't mix w/ everything

Problems Molarity $M = \text{mol/L}$

1) 3 Molar

1) 3 Molar
2) $EM = \text{mol} / 3 = 15 \text{ moles}$

$$3) 10M = 100/L = 10L$$

What's the difference between Homogeneous and Heterogeneous?

what's something
micible?

what's something invincible?