

Write the formulas for the following chemical compounds:

- 11) silicon dioxide SiO_2
- 12) nickel (III) sulfide Ni_2S_3
- 13) manganese (II) phosphate $\text{Mn}(\text{PO}_4)_2$
- 14) silver acetate $\text{Ag}_2\text{C}_2\text{O}_4$
- 15) diboron tetrabromide B_2Br_4
- 16) magnesium sulfate heptahydrate $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$ (skip.)
- 17) potassium carbonate K_2CO_3
- 18) ammonium oxide $(\text{NH}_4)_2\text{O}$
- 19) tin (IV) selenide SnSe_2
- 20) carbon tetrachloride CCl_4

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