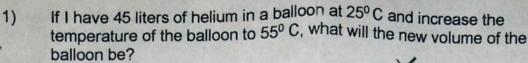
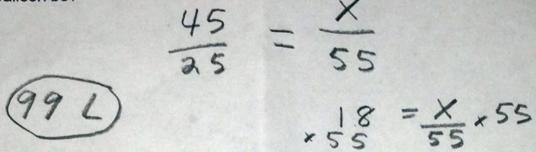
## Charles's Law





Calcium carbonate decomposes at 1200° C to form carbon dioxide and 2) calcium oxide. If 25 liters of carbon dioxide are collected at 1200° C, what will the volume of this gas be after it cools to 250 C?

TROO = 25esale.cozuz Notesale.cozuz 130 liters of a Diage

3) O leve 130 liters of grain at a temperature of 250° C. If I cool the gas until the volume decreases to 85 liters, what will temperature of the gas be?

