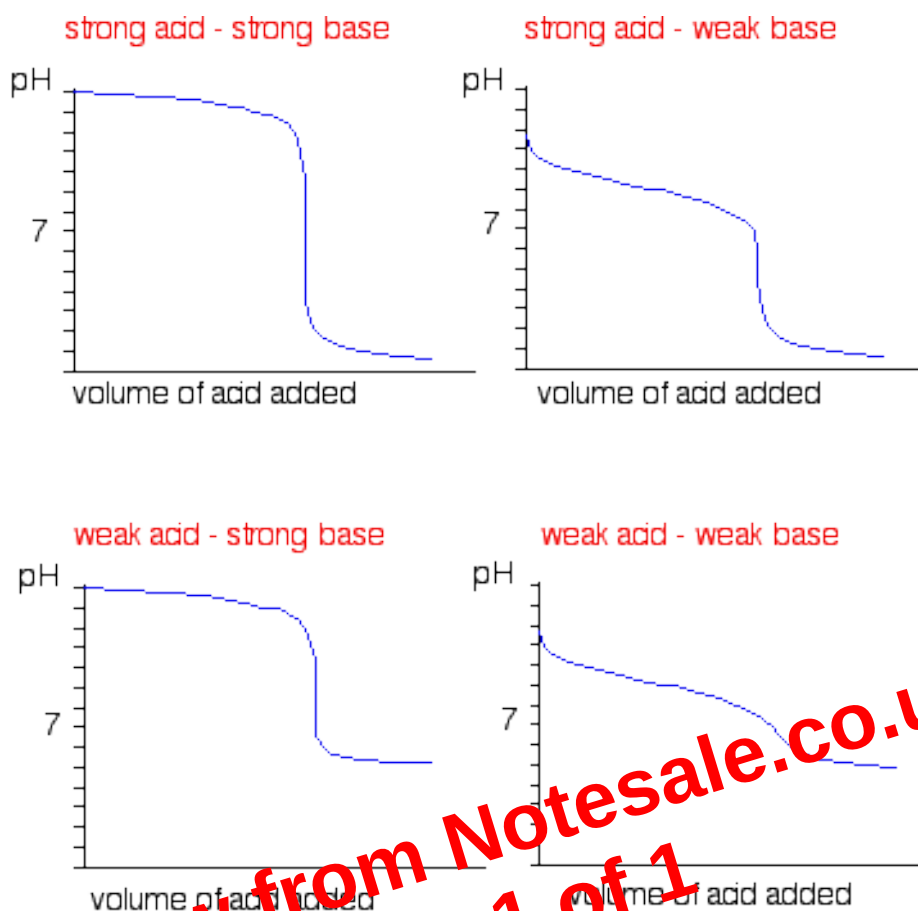


pH Curves



Indicators which change colour within the pH range of the vertical section of the graph will be chosen for that specific experiment

When the indicator changes colour this is known as the **end point**

The point in a titration at which the volume of one solution has reacted exactly with the volume of the second solution. That matches the stoichiometry of the reaction that is taking place, this is known as the **equivalence point**. (usually located in the middle of the vertical section)