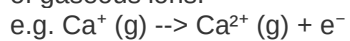


Once an electron has been removed from an atom, remaining electrons pulled closer to nucleus and more tightly held.

As a result, successive ionisation energies show increase in value.

Second ionisation energy is the energy required to remove a second electron from each ion in a mole of gaseous ions.



Second ionisation energy value always greater than first ionisation energy.

Preview from Notesale.co.uk
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