4BP 1LP		seesaw
	X AX AX ₄ E ₁ Sawhorse or Seesaw	
3BP 2LP	F = 90° F = 120°	T-shape
6ВР	F F F	Octahedral
5BP 1LP	Final Sole Control of 11 Preview Page 6 Preview Page 7 Prev	Square (vramidal
4BP 2LP	Previe Page F 90° III	Square

E.g. H_2O

H: 2* (1 electron)
O: 6 electrons

Total= 8 electrons

 $2\ bonds$, seen from H_2O , no double bonds as Hydrogen isn't -vely charged

2BP = 4 electrons (need a total of 8 electrons) so add 2LP = 8 electrons

2BP 2LP = bent

