

## Lecture 1

### Elements

### 1.3 - Classification of Matter

8/30/17

- cannot be broken down into other elements
- 118 known elements
- own symbol
- own properties
- around 132 million compounds + more everyday

"a specific instance  
↑ of matter"

substance pure or mixture? element? compound or homo or heterogenous?

copper pure element homo

apple juice mixture mixture of cmpd homo

octane-C<sub>8</sub>H<sub>18</sub> pure compound homo

chicken soup mixture mixture of cmpd hetero

### Solids

- ① Crystalline solids - atoms or molecules are in a regular pattern  
fixed positions, don't move (vibrate in place)

eg - diamond

- ② Amorphous solids - no regular pattern; random arrangement of atoms & molecules  
eg - charcoal

Liquids - atoms/molecules not packed together tightly (pack about as closely as in solid)  
free to move about

fixed volume, assume shape of the container

Gases - lot of space b/w atoms/molecules → allows compression

random movement - high velocity

assume shape AND volume of the container

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