Monosaccharides: Simple Sugars (one).

Disaccharides: 'Double Sugars' (two monosaccharides)

Polysaccharide: large molecules formed from many monosaccharides.

As polysaccharides are very large molecules, they are insoluble. This makes them good for storage.

Benedict's Test: B Solution su reducing sugar is to ate Pyill 1 50 soluble red precipitate of copper oxide. B reagent turns from blue -> red with reducing sugar.

Non-Reducing sugars don't change colour of B Reagent. Need to add hydrochloric acid —> Heat (hydrolysing the disaccharide into it's monosaccharides) -> Add sodium hydrocarbonate (neutralise the acid) —> re-test the B Solution.

OH = Hydroxyl.

When monosaccharides join, a molecule of water is removed = Condensation Reaction. A Glycosidic Bond is formed.

When water is added to a disaccharide it breaks the Glycosidic bond = releasing monosaccharides. = Hydrolysis Reaction.

3 Monosaccharides: Glucose / Fructose / Galactose

3 Disaccharides:

Lactose: Glucose + Fructose (Table Sugar)

Glucose is a hexose

sugar (5 Carbons/ 1

Oxygen) / 6 Atom Fig. O Feducose)

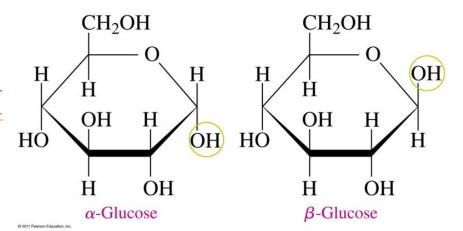
Sulfate. When

Carbohydrates

Respiration:

Oxygen + Glucose -> CO2 + Water

6O2 + C6O12H6 --> 6CO2 + 6H2O



Glucose / Fructose = Structural isomers: have the same molecular formula (C6H12O6) / different structural formula.

Functions of Carbohydrates:

- 1. Substrate for respiration (glucose is essential for cardiac tissues).
- 2. Energy stores (e.g. starch, glycogen).
- Structural (e.g. cellulose, chitin in arthropod exo-3. skeletons and fungal walls).
- Transport (e.g. sucrose is transported in the phlo-4. em of a plant).
- Recognition of molecules outside a cell (e.g. 5. attached to proteins or lipids on cell surface membrane).

Carbohydrates Contain:

Carbon (C)

Hydrogen (H)

Oxygen (O)

The 5-carbon monosaccharide ribose is an important component of coenzymes (ATP) and the backbone of the genetic molecule known as RNA.