

- Enzymes are proteins with a long chain of amino acids and very specific three-dimensional shape
- They have an area that is designed to match a specific molecule known as *substrate*
 - This area is called *active site*
 - Analogy for enzyme-substrate activity is a lock and key
 - The lock represents the active site and the key the substrate
 - Because the three-dimensional shape of the internal portion of the lock is complex and specific only one key will fit
- The substrate must enter with a minimum rate of motion that will provide enough energy for the reaction to occur
 - This energy is called *activation energy* and it is not provided by the enzymes
 - Enzymes just lower the minimum energy that is required
- Enzymes are not reactants and therefore they are not used up in the reaction
 - Enzymes act like catalysts and they can be used many times
 - A *catalyst* increases the rate of reaction without being consumed
- An enzyme cannot force a reaction that would not happen without the enzyme's presence to occur
- The reaction may be more likely to occur with an enzyme present, as the activation energy will be lower

Enzyme-catalysed reactions – factors

- All chemical reactions are fundamentally molecules colliding
 - At a high enough rate of speed, if the molecules are capable of reacting with each other, there is a chance that a reaction will occur
 - Enzymes cannot change these fundamentals
- There are many factors that affect the rate of an enzyme-catalyzed reaction
 - Temperature
 - The rate of the motions of the enzyme and the substrate are dependent on the temperature of the fluid in which they move
 - The higher the temperature of the fluid, the faster the molecules move and the faster the molecules move, the more often they collide
 - Reactions with or without enzymes will increase their reaction rate as the temperature rises
 - Reactions that use enzymes have an upper limit, which is based on the temperature at which the enzyme begins to lose its shape