

• Group 2

+8	4	Be	बेरील	=	2, 2
	10	Mg	मैगनेशियम	=	2, 8, 2
+8	20	Ca	कैल्शियम	=	2, 8, 8, 2
+18	38	Sr	स्ट्रॉन्शियम	=	2
+18	56	Ba	बारियम	=	2
+82	88	Ra	(Radioactive) रेडियम	=	2

Group 2, (II A)

Alkaline earth metals

Valence $e^- = 2$

Valency = 2

• Group 13

(For p-block - ~~Magnesium~~ Magnesium is in Group 2, 13, 14, 15, 16, 17, 18, 32)

8	5	B	बोरॉन	=	2, 3
18	13	Al	अलुमिनियम	=	3
18	31	Ga	गैलियम	=	3
54	83	In	इंडियम	=	3
86	113	Tl	थैलियम	=	3

Group 13 (III A)

Boron family

Valence $e^- = 3$

Valency = 3