

LAB 1: CARBOHYDRATES

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$$\frac{A_b}{c_b} = \frac{A_d}{c_d \left(\frac{0.1}{0.6}\right)}$$

$$m_1 = c_d(V_1) :$$

$$\text{Variables: } A_c = 0.456$$

$$c_c = 0.1 \text{ mg/mL}$$

$$A_d(c_b) = A_c \left(\frac{0.1}{0.6}\right)(c_d)$$

$$m_1 = 0.87 \frac{\text{mg}}{\text{mL}}(10 \text{ mL})$$

$$A_d = 0.663,$$

$$m_1 = 8.7 \text{ mg}$$

$$V_1 = 10 \text{ mL}$$

$$c_d = \frac{(A_d)(c_c)}{A_c \left(\frac{0.1}{0.6}\right)}$$

$$c_d = \frac{(0.663)(0.1)}{0.456 \left(\frac{0.1}{0.6}\right)}$$

$$c_d = 0.87 \frac{\text{mg}}{\text{mL}} = 0.87 \frac{\text{g}}{\text{L}}$$

At $t=0$ min, the concentration and mass of starch in glucose/starch mixture is 0.87 g/L and 8.7 mg.

Comment [S5]: Missing: data analysis
0/4

R2. Estimating glucose concentration at the equilibrium

$$c_{eq} = \frac{(c_o)(\alpha)}{1+\alpha}$$

$$\text{Variables: } c_o = 48.3 \text{ mg/mL}$$

$$V_1 = 10 \text{ mL}$$

$$c_{eq} = \frac{(c_o)(V_1)}{V_1 + V_2}$$

$$V_2 = 200 \text{ mL}$$

$$c_{eq} = \frac{\left(48.3 \frac{\text{mg}}{\text{mL}}\right)(10 \text{ mL})}{10 \text{ mL} + 200 \text{ mL}}$$

$$c_{eq} = 2.3 \frac{\text{mg}}{\text{mL}} = 2.3 \frac{\text{g}}{\text{L}}$$

The concentration at equilibrium is 2.3 g/L.

Comment [S6]: 7/7

R3. Estimating the diffusion coefficient D (cm²/min) of glucose across the dialysis membrane