7.	The reagents used in Williamson's synthesis for the preparation of diethyl ether are	
	(a)	sodium ethoxide and methyl bromide
	• •	sodium ethoxide and ethyl bromide
		sodium methoxide and methyl bromide
	(d)	sodium methoxide and ethyl bromide
8.	The	boiling point of ethers are
	<u>(a)</u>	lower than that of isomeric alcohols
	(b)	equal to that of isomeric alcohols
	(c)	greater than that isomeric alcohols
	(d)	equal to that of alkanes
9.	Eth	ers are soluble in
		10 CO.U.
	(a)	hydrocarbons (b) goldetrachloride
	(c)	benzene NOTO all
10.	The	hydrocarbons benzene  (b) concerrachloride benzene  (c) all  reación of ethers areducto the presence of  alkyl groups lone pair of electrons on oxygen  (d) all
D	97	vio pago
	(a)	alkyl groups (b) C - O bonds
	(c)	lone pair of electrons on oxygen (d) all
11.		dichlorodiethyl ether is formed when
	(a)	dimethyl ether reacts with chlorine in dark
	` ′	diethyl ether reacts with chlorine in dark
		diethyl ether reacts with chlorine in the presence of sunlight
		dimethyl ether reacts with chlorine in dark
12.	$C_2$ H	$H_5 - O - C_2 H_5 + H_2 O \xrightarrow{ dil   }$ . The product is
	(a)	$C_2 H_5 OCH_3$ (b) $H_2 O_2$
	` ′	$\begin{array}{ccc} 2 \ C_2 \ H_5 \ OH \end{array} \tag{d}  C_2 \ H_5 \ H_2 \ SO_4$