

- 20.8 torr (Clausius-Clapeyron Equation)
- What mass (in g) of NH_3 must be dissolved in 475 g of methanol to make a 0.250 m solution?
 - 2.02 g NH_3
- A solution is prepared by dissolving 15.0 g of NH_3 in 250.0 g of water. The density of the resulting solution is 0.974 g/mL. The molarity of NH_3 in the solution is
 - 3.24 M

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