

RATES OF CHEMICAL REACTIONS

- ✗ Some chemical reaction occurs instantaneously, think of the speed of an explosion, Others weathering of rocks, occurs extremely slow, Ripening of fruits, fire works.

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- ✗ During a chemical reaction, the concentration of the reactant decrease, while the concentration of the product increase. We can therefore measure the rate of a reaction by monitoring changes in the concentration of reactants or products

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- ✗ Able to measure
- ✗ The change in volume of a gas
- ✗ The rate of formation of precipitate (ppt)
- ✗ The change in pH where the pH of reactants and products differ.

RATES OF CHEMICAL REACTIONS

✖ The effect of Catalyst on reaction rates

A catalyst is a substance that speeds up a chemical reaction, but its not used up in the reaction.

A catalyst do not appear in the chemical equation.

Take part in the rxn, but are unchanged, chemically, at the end, though solid catalyst may undergo physical changes.

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- ✗ The effect of Catalyst on reaction rates
- ✗ Catalyst provide different or easier way to form products.
- ✗ It can provide a surface the reactants to react
- ✗ It can provide an alternative route for the reaction to take place.
- ✗ Catalyst may bring reacting molecule closer together- thus more opportunities for reactant molecules to collide.
- ✗ Note that different catalyst have different effects on reaction rates