

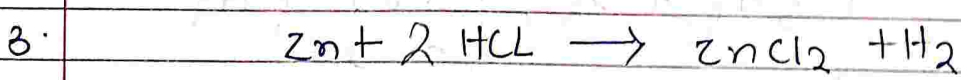
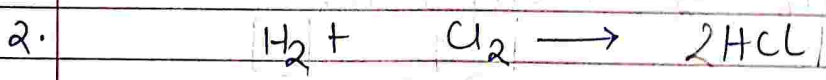
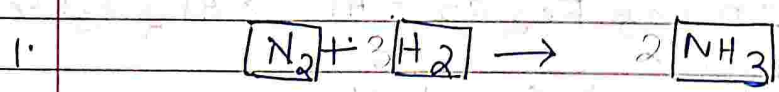
Equations

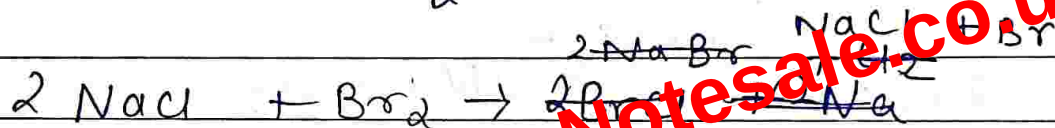
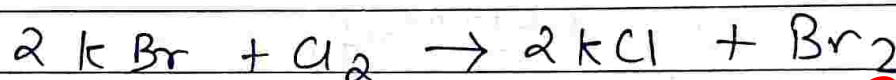
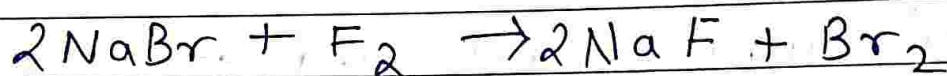
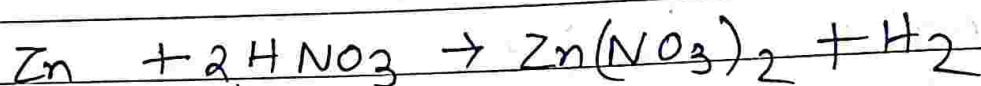
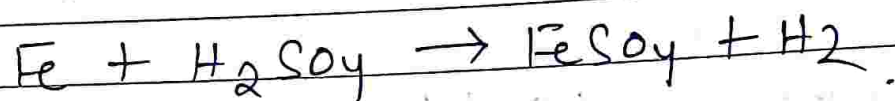
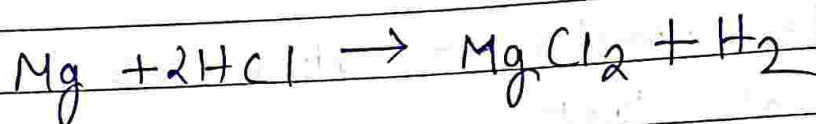
- Chemical equations are statements which represent a chemical reaction.
- In a chemical equation there is a LHS called reactants and a RHS called products
- The chemical equation has to be balanced. This means that the number of atoms on the LHS side should be equal to number of atoms on the RHS.
- This is followed because the law of conservation states that matter cannot be created or destroyed.

Preview from Notesale.co.uk
page 3 of 14



Balancing Chemical eqns.





4) Double Displacement

