Pago No.:	1- 1	· · · · · · · · · · · · · · · · · · ·
EST.		Page tio
		Equations
i vi	-	
	7	Chemical equations are statements which
74		represent a chemical reaction.
F21 1	<b>→</b>	Tro a close of the second of t
λ) <sup>σ</sup>		In a chemical equation there is a LHS called products
.517		called reactants and a parts called products
+ 1	$\rightarrow$	The chemical equation has to be balanced
B-B		This means that the number of atoms
P. A.		on the little order observed
	-	number of atoms on the RHS.
ur I	-	Two to the CO.Y.
		This is tollowed to save the law of conservation
		State that mater cannot be created or
Y.		number of atoms on the RHS.  This is followed a Gaze the law of conservation States that Main connot be created or destroyou.  HARRY ALL ALL CANNOT be created or
	orey	HOACE -> 2 HCI
	0	
	40	Balancing Chemical eque.
,	1 1	
	1.	(N2+3H2) -> 2 NH3
		A self to real to the self to
	٦.	H2+ Cl2 -> 2HCL
	2 .	zn+2HCL -> znc12+H2
	3·	Ln+ / ITCL -> Zncl2 +112
	4.	Py+ < 02 -> 2P,06
		2 56

Page No .: Mg + 2HC1 -> Mg C12 + H2 Fe + H2 Soy -> Fe Soy + H2 In +2HN03 + Zn(NO3)2 + H2 2KI+U2 -> KCI+I2 2NaBr. + F2 ->2NaF+Br2 & CaI2+RU2 -> &CaU2 +RI2 2 KBr + Cla -> 2 KCl + Brz 2 Nacl + Brz -> 2 Brote Salva 2 Helie W -> 2 Brote Salva 2 Helie W -> 2 Brote Salva Preview -> 2 Brote Salva Preview -> 2 Brote Salva -> 2 Brote Salva -> 2 Helie W -> 2 Brote Salva -> 3 Brote Salva -> 4 Brote Salva -> 5 Brote Salva -> 4 Brote Salva -> 5 Brote S & Double Displacement  $A/B + C/D \rightarrow AD + CB$ Na/oH + HICI -> Nacl + H20 Boleia + Nazboy -> Basoy +2Wac/ X (ag) (ag) (s)/ (ag)

(orite)/

pre l'plitation reaction.