UNIVERSITY OF LAGOS

DEPARTMENT OF CHEMISTRY

B.Sc./ B.Sc. (Edu) Hons. Examination, 1st Semester 2017/2018

CHM 213: BASIC ORGANIC CHEMISTRY

Attempt All Ouestions

Total Time: 21/2 Hour

This examination is in two parts; part A for 11/2 hr after which it will be collected

Part B is for only 1 hr.

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Department:

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PART A

- The reagent used in the reduction of ketone in the preparation of tertiary (3°) alcohol is LiAJH4(b) NaBH4 (c) RMgCI/H+ (d) All of the above.
- In the dehydration reaction of CH3CH2CHCH2OH in the presence of sulphuric acid, at 140°C. The product formed is?

- alcohol
- The product of the reaction of metal alkoxides and primary alkyl halides is
- alkene

ether

- alcohol
- 5. The name of the organic compound: CH₃CH(CH₂I) CH₂ CHO is
- 4-iodo-3-methyl propanal
- 3-iodomethyl butanal
- 3-iodomethyl-3-methyl propanal
- 4-iodo-3-methyl butanal
- The product of the reaction of acid chloride and aromatic hydrocarbon in the presence of a Lewis catalyst is

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- ketone (b)
- aldehyde
- ester
- alcohol

- 7. Which of the following compounds would you expect to possess the highest boiling point
 - (a) CH3CH2CH2CHO (b) CH3CH2OCH2CH3 CH3CH2CH2CH2OH
- (c) CH3CH2CH2COOH (d)
- Which of the following would be oxidised to propan-2-one
 - (a) butan-1-ol (b) ethanol (c) 2-methylpropan-2-ol
- (d) propan-2-ol

- Which of the following statements are correct:
 - i. 3° alcohols have lower boiling points than 1° alcohols with an similar molecular weights
 - ii. Alcohols undergo nucleophilic substitution
 - iii.3° alcohols undergo dehydration more readily than 1° alcohols
 - iv. 3° alcohols are oxidised to ketones
 - (a) i, ii, iii & iv
- (b) i, ii & iii (c) i, iii & iv
- (d) ii. iii & iv
- Choose the most acidic compound from the following:









- Which of the following statements is incorrect:
 - (a) Ethers are flammable
 - (b) Ethers form peroxides by free radical oxidation
 - (c) Ethers are widely used as extraction solvents
 - (d) An ether oxygen can only be found within an acyclic carbon chain
- How many 1.2-disubstituted benzene is(are) possible
 - (a) 1

The rate of SN2 reaction is

$$(a) = k \left[RX \right]^{2} (b) = k \left[RX \right] \left[Nu \right] (c) = k \left[RX \right] (d) = k \left[RX \right]^{2} \left[Nu \right] (e) = \left[RX \right] \left[Nu \right]$$

Give the IUPAC nomenclature for the compound below:

- (a) 1.4-ketohexanol
- (b) 4-oxocyclohexanol
- (c) 4-oxohexanol
- (d) 4-hydroxycyclohexanone
- Which of the following would be oxidised to propan-2-one
 - (a) butan-1-ol (b) ethanol
- (c) 2-methylpropan-2-ol
- (d) propan-2-ol