

IN CASE YOU FORGOT

for positive ions:-
reactivity series

for negative ions:-
reactivity series

Police
Sergeant
Lee
Charlie
M
A
C
Z
I
N
T
L
H

Potassium
Sodium
Lithium
Calcium
Magnesium
Aluminium
CARBON
Zinc
Iron
Nickel
TIN
Lead

HYDROGEN

caught
me
stealing
gold
plates

copper
mercury
silver
gold
platinum

above hydrogen, ↑
reactive so hydrogen
will be released instead.

carbon can't
reduce metals
above it cuz they
are ↑ reactive.

below hydrogen, ↓ reactive
so metal will be released
instead.

SO_4^{-2}
 NO_3^{-1}
 OH^{-}

halides
released
instead of OH.
(↑ reactive)

Cl^{-}
 Br^{-}
 I^{-}

OH released
instead of
negative ions
(↓ reactive)

soon
no
one
can
bring
ice

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