

Structure of atom

Note :.

2- Dalton's atomic theory proposed that matter is composed of small individual particles called atoms .

2. Atoms are further compared into fundamental particle at 9.c. election , proton and neutron

3- Election was discovered by cathode ray discharge tube experiment

4. Nucleus was discovered by Rutherford

in 1911 .

ves Explain the cathode ray discharge tube experiment .

A cathode ray discharge tube is made up of glass with two electrodes at very low pressure at high voltage , current starts flowing through a stream of particles moving in the tube from cathode to anode . These were called cathode rays . When the perforated anode is taken the cathode ray stuck the

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Page 1 of 1