

PROPERTIES OF ACIDS

{1}- ACIDS HAVE A SOUR TASTE.

{2}- ACIDS TURNS BLUE LITMUS TO RED LITMUS.

{3}- ACIDS DON'T FEEL SOPHY IN TOUCH.

{4}- ACIDS SOL[^] CONDUCT ELECTRICITY (THERE IONS IN FREE STATE).

{5}- ACIDS REACT WITH METAL TO FORM HYDROGEN GAS.



{6}- ACIDS REACT WITH METAL CARBONATE (AND METAL HYDROGEN CARBONATE) TO FORM CARBON DIOXIDE GAS.



{7}- CARBON DIOXIDE GAS REACT WITH LIME WATER (CALCIUM HYDROXIDE SOL[^]) AS FOLLOWS.

{8}- ACIDS REACT BASES (OR ALKALIS) TO FORM SALT AND WATER.



IT ALSO CALLED NEUTRALIZATION.

{9}- ACIDS REACT METAL OXIDE TO FORM SALT AND WATER.

