- b) **Reaction with Sodium Hydroxide**: Formation of phenols.
- c) **Reaction with Sodium**: Formation of biphenyls (Wurtz-Fittig reaction).
- Important Name Reactions:
 - a) **Fittig Reaction**: Formation of biphenyl from haloarenes and sodium.
 - b) **Gattermann-Koch Reaction**: Forming aromatic aldehydes from haloarenes.
- 3. Important Conversions:
- **Haloalkane to Alcohol**: Use **KOH (ag)** for nucleophilic substitution.
- **Haloalkane to Alkene**: Use **Alcoholic KOH** for elimination reaction (E2 mechanism).
- **Haloalkane to Amine**: Use **Ammonia (NH3)** under pressure for nucleophilic substitution.
- **Haloarene to Phenol**: Use **Sodium Hydroxide (NaOH)** to replace halogen with OH group.

- **Haloarene to Biphenyl**: Use **Sodium metal (Na)** in dry ether (Murchard reaction).

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