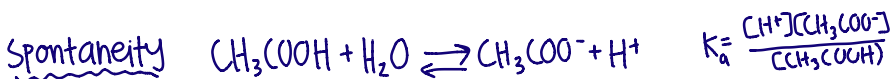




$$Q = \frac{[\text{Zn}^{2+}]\text{PH}_2}{[\text{H}^+]^2}$$

More active = wants to be oxidized
Least active = wants to be reduced

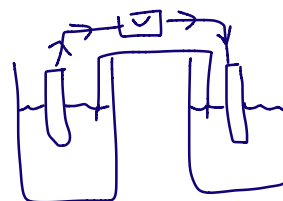


(Le Chatelier's Principle)



forward reaction is spontaneous

Add electric current (Electrolysis)



Preview from Notesale.co.uk
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