

SAMPLE QUESTION PAPER CHEMISTRY CLASS XII

Design of Question paper

Time: 3 hrs. Maximum Marks:70

A. Weightage to different forms of questions

S. No.	Type of question	Marks per Question	Total number of Questions	Total marks	
1	VSA	1	8	8	
2	SAI	2	10	20	
3	SA II/Value Based	3	9	27	.co.u
	Question		Note	25an	
4	LA	5	W 134	110	
Total		W TO	34 0	70	
	-116	5 //	40	1	<u></u>

B. Typology of Questions

S. Typology Weightage in marks percentage 1 Knowledge Based 14 20% Conceptual 21 30%	,
1 Knowledge Based 14 20%	age
1 Knowledge Based 14 20%	
Conceptual 21 30%	
Conceptual 21 30%	
21 0070	
2 Understanding	
3 Inferential Type 14 20%	
4 Reasoning Based 11 15%	
5 Skill Based 10 15%	
Total 70 100%	

1



- 1/2 21 (a) Acidic flux is SiO₂
 - 1/2 Basic flux is CaO

(b)

(i) Cu₂O undergoes self reduction to form blister copper as

$$2Cu_2O + Cu_2S \longrightarrow 6Cu + SO_2$$

(ii)
$$Fe_2O_3 + 3C \longrightarrow 3CO + 2Fe$$

OR

- (a) Cryolite reduces melting point of alumina.
- (b) Concentration of ore is carried out by digesting the powdered ore with a concentrated solution of NaOH at 473-523 K and 35 -36 bar pressure. Al₂O₃ is leached out as sodium of the luminate and SiO₂ as sodium silicate leaving behind impurities.

 Al₂O₃ (s) + 2 NaOH(aq) + 3H₂O \rightarrow 2Nr[1]OH)₄] (aq)

 The resulting solution is filtered, collected 3.d. Feutralised by passing CO₂ gas through it. Then hydrated Al₂O₃

$$Al_2O_3(s) + 2 NaOH(aq) + 3H_2O \rightarrow 2Nn[NOH]_4] (aq)$$
 1

gets precipitated leaving sodium silicate in the solution.

$$2Na[Al(OH)_4] (aq) + CO_2(g) \longrightarrow Al_2O_3.xH_2O(s) + 2NaHCO_3(aq)$$
 1/2

Sodium silicates remain in the solution and hydrated alumina is filtered, washed, dried and heated to get pure alumina (Al_2O_3) .

$$Al_2O_3 .x H_2O(s) \longrightarrow Al_2O_{3(s)} + x H_2O(g)$$
 1/2

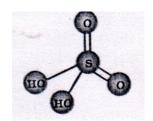
22.(i)
$$H_3PO_3 + 2AgNO_3 + H_2O \longrightarrow 2Ag + HNO_3 + 2H_3PO_4$$
 1

(ii)
$$3Cl_2 +6 NaOH$$
 (conc.) \longrightarrow $5NaCl + NaClO_3 +3 H_2O$ 1

(iii)
$$2XeF_2 + 2H_2O$$
 \longrightarrow $2Xe + O_2 + 4H^+ + 4F^-$ 1



23(i)



1

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(ii) Since the formation of ozone from oxygen is endothermic reactions (e)	ectric discharge prevents its
(iii) Since the formation of ozone from oxygen is endothermic reaction shere elected decomposition. (iii) Bleaching by sulphy Course is temporary because or its reducing nature.	
(iii) Bleaching by sulphy coolede is temporary because of its reducing nature.	1
24 (i) Amylose is a long unbranched chain polymer α -D(+) glucose.	1/2
Amylopectin is a branched chain polymer of $lpha$ -D glucose.	1/2
(ii) Globular protein and its shape is spherical.	1/2 +1/2
(iii) Liver and adipose tissue.	½ +1/2
25. (i) Antiseptics are the chemicals applied to the living tissues either to kill or pricroorganisms .e.g dettol.	orevent the growth of ½ +1/2
(ii) Antioxidants are the compounds which retard the action of oxygen on food ar decomposition by oxidation.e.g BHA	nd reduces its rate of ½+1/2