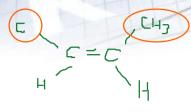
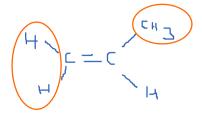
E-Z isomers

.7 December 2012 .7:18 E isomers- Entgegen: opposites Z isomers- Zusammen: same



H (CH3)



- Priority groups (those with the greatest Mr) are on the same side of the double bond
- Therefore they are <u>Z isomers</u>
- The old name for it would be Cis isomer
- Priority groups are on opposite ends of the double bond
- This makes them E isomers
- Old name would be Trans isomers
- There is <u>no distinct priority group</u> so <u>no isomer</u> is formed

Geometric isomers arise as a result of restricted rotation around the double bond at room temperature

Cis/trans system breaks down due to no priority group being the same

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