Basic chemical anthocyanin<sup>4</sup> 
$$R^{7}$$
  $R^{8}$   $R^{5}$   $R^{5}$  structure of  $R^{7}$   $R^{6}$   $R^{5}$ 

Conclusion

sale.co.uk A pH scale was established for tive indicator. Based on the pH used to discover the pH level of scale established, the natural in dicators are then The A level of vinegar, orange juice, bleach, rough estimation balling soda, grapefruit juick lemon, and egg whites were determined. The pH level of bleach cannot be determined as it is not in the range of the natural indicators. Natural indicators are one of the ways to determine the pH level of household items. It is affordable and safe to be conducted in home environment.

## Strength

1. It is possible to determine the approximate pH of a solution with the range from 1-14 as it displays variety of colours.

## Limitations

1. It is difficult to determine estimated pH range of acidic substances due to only one indicator that can be use, which is red cabbage.

<sup>&</sup>lt;sup>4</sup> (n.d.). Retrieved February 21, 2016, from https://en.wikipedia.org/wiki/Anthocyanin#/media/File:Anthocyanidine.svg